

Write the correct name or chemical formula for the following compounds:

| | | Ionic or Molecular |
|---|-------|--------------------|
| a. CaCl_2 | _____ | _____ |
| b. Copper (II) phosphate | _____ | _____ |
| c. SO_3 | _____ | _____ |
| d. Zinc Carbonate | _____ | _____ |
| e. Nickel (II) Chloride hexahydrate | _____ | _____ |
| f. Cadmium Bromide | _____ | _____ |
| g. H_2O_2 | _____ | _____ |
| h. Li_2CO_3 | _____ | _____ |
| i. $\text{Au}(\text{NO}_3)_3$ | _____ | _____ |
| j. Dinitrogen trioxide | _____ | _____ |
| k. $\text{K}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$ | _____ | _____ |
| l. Ammonium Nitrate | _____ | _____ |
| m. Magnesium Hydride | _____ | _____ |
| n. Manganese (IV) oxide | _____ | _____ |
| o. Beryllium hydroxide | _____ | _____ |