

Name \_\_\_\_\_ Site \_\_\_\_\_

Provide the name or formula for each of the following:

Name:	Formula
1. copper (II) sulphate pentahydrate	
2.	$\text{MgSO}_4 \cdot 7 \text{H}_2\text{O}$
3. potassium carbonate octahydrate	
4.	$\text{MgCl}_2 \cdot 6 \text{H}_2\text{O}$
5. barium chloride dehydrate	
6.	$\text{Cd}(\text{NO}_3)_2 \cdot 4 \text{H}_2\text{O}$
7. Lithium bromide trihydrate	
8.	$\text{Na}_2\text{S}_2\text{O}_3 \cdot 5 \text{H}_2\text{O}$
9. cobalt (II) chloride hexahydrate	
10.	$\text{AlCl}_3 \cdot 6 \text{H}_2\text{O}$
11. zinc sulphate nonahydrate	
12.	$\text{CaCl}_2 \cdot 2 \text{H}_2\text{O}$
13. barium hydroxide monohydrate	
14.	$\text{Na}_2\text{SO}_4 \cdot 10 \text{H}_2\text{O}$
15. magnesium silicate pentahydrate	